PRELIMINARY NOTE

Diazotisation of 4-Aminotetrafluoropyridine in a Sulphuric Acid-Acetic Acid-Propionic Acid Medium

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SUMMARY

4-Aminotetrafluoropyridine can be diazotised successfully by adding it to a solution of sodium nitrite in a mixture of 98% sulphuric acid, acetic acid and propionic acid; addition of mesitylene, <u>m</u>-cresol, 3,5-dimethylphenol or 2-naphthol to the solution of tetrafluoro-4-pyridyldiazonium ion thus produced provides the expected azo-compounds in good yields.

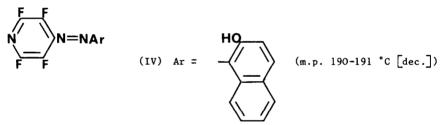
The recent disclosure [1] of a diazotisation technique for pentafluoroaniline that avoids use of hydrofluoric acid prompts us to report that we have made similar progress during our studies on coupling reactions of tetrafluoro-4-pyridyldiazonium ion [2].

In a typical diazotisation-coupling reaction, an 86:14 v/v mixture (20 cm^3) of acetic and propionic acid was added dropwise to a stirred solution of dry sodium nitrite (0.7 g, 10 mmol) in 98% sulphuric acid (20 cm^3) kept at < 30 °C by means of an ice bath; the nitrosating medium

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thus produced was cooled to <u>ca</u>. 0 °C, treated with 4-aminotetrafluoropyridine (1.65 g, 10 mmol), then stirred at 0 °C for 3 hours (possibly an excessive period) before mesitylene (1.2 g, 10 mmol) was added dropwise. Immediately a red precipitate appeared, and after 3 more hours this was recovered (the reaction mixture was poured into water first) and purified by dry-column flash chromatography to provide the known [2] 4-(2,4,6-trimethylphenylazo)-2,3,5,6-tetrafluoropyridine (I) in 77% yield (2.3 g, 7.7 mmol). Similarly, the new azo-compounds (II)-(IV) were prepared by using <u>m</u>-cresol (yield 66%), 3,5-dimethylphenol (89%), or 2-naphthol (69%), respectively, as the coupling component instead of mesitylene. Use of concentrated sulphuric acid in conjunction with aliphatic carboxylic acids as a diazotisation medium for weakly-basic arylamines is a well-established technique [3].

(I) Ar =
$$2,4,6-Me_3C_6H_2$$
 (m.p. $142-143$ °C)
(II) Ar = $4-HOC_6H_3Me-2$ (m.p. $172-173$ °C)
(III) Ar = $4-HOC_6H_3Me_2-2,6$ (m.p. $156-158$ °C)



The new azo-compounds were identified by elemental analysis (C,H,F,N) and spectroscopic methods [i.r., mass, and n.m.r. $({}^{1}H, {}^{1}9F)$]; the mesitylene derivative (I) has been shown by <u>X</u>-ray crystallography [4] to prefer to exist in the <u>trans</u>-form, and the close correspondence between ${}^{19}F$ n.m.r. chemical shift data for members of the series (I)-(IV) indicates that the same is true for the other azo-compounds. Even in Manchester sunlight, <u>trans</u>-4-(2,4,6-trimethylphenylazo)tetrafluoropyridine (I) dissolved in deuteriochloroform isomerizes partly to the <u>cis</u>-form; after irradiation with u.v. light (through soda glass), a photostationary state was reached after 20 hours, the <u>cis</u>:trans ratio being 1:3 (establishment of equilibrium was followed by ${}^{1}H$ n.m.r.). <u>Cis</u>-trans photoisomerization of azo-compounds (II)-(IV) is under investigation, as is the possible occurrence of azo-hydrazone tautomerism. We are greatly indebted to the SERC for awarding a CASE Studentship to A.C.A.

- 1 H. Iwamoto, T. Sonoda, and H. Kobayashi, J. Fluorine Chem., <u>24</u> (1984) 535.
- 2 R.E. Banks, A.R. Thompson, and H.S. Vellis, J. Fluorine Chem., <u>22</u> (1983) 499.
- 3 K.H. Saunders, 'The Aromatic Diazo-compounds and their Technical Applications', Arnold, London, 1936, p.11.
- 4 R.G. Pritchard, Personal Communication.